

# Recycle of Zincates and Aluminum to Fed Zn-Air Fuel Cells †

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**Abstract:** It is well known that the Zn-air battery depth of discharge is mainly governed by the electrochemistry of zinc anode in concentrated alkaline solutions leading to undesired precipitation of insulating zinc oxide at the electrode surface due to progressive increase in zincate concentration. Among the different approaches adopted to overcome this problem, mechanically and electrically rechargeable as well as refuelable systems have been deeply studied for automotive and stationary applications. In a refuelable Zn-air battery, the alkaline aqueous electrolyte is pumped into the cell, either carrying Zn electroactive fuel or flowing through a packed bed of Zn particles. In a previous paper, we adopted a mechanically refuelable tapered-end flow Zn-air fuel cell with Zn micro-spheres, and we studied the effect of electrolyte aging on the behavior of the Zn anode during battery discharge. The results have shown that the cell potential decrease in the battery discharge curve is mainly due to the anodic overpotential increase, as evidenced by means of EIS (Electrochemical Impedance Spectroscopy) measurements. Given that this effect can be attributed to the physicochemical modifications induced by the Zn passivation occurring at higher zincates concentration, the performance of the battery system can be notably improved by a relevant regeneration of spent alkaline zincate bath.

Here, we propose a novel method to recover zinc from alkaline zincate baths by using aluminum electrodes. Aluminum metal and zincate ions give place to a single-displacement reaction producing a layer of zinc on the Al surface. The deposits of zinc metal on aluminum foils have been studied by means of X-ray diffraction (XRD), scanning electron microscopy (SEM), and the electrochemical process has been followed by open circuit measurements in order to put the basis of future developments.

**Keywords:** zinc air fuel cell; displacement reaction; zincate regeneration.

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