

# Insight into the Effect of Iron Alumina Catalyst on Methane Decomposition in Terms of CNT Quality and Methane Conversion in the Fluidized Bed Reactor †

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**Abstract:** The need for clean, sustainable energy sources increases due to environmental fouling from fossil fuel emissions. Hydrogen is one of the best clean fuels with high energy density and no adverse environmental effects. The most common hydrogen production method, i.e., Steam Methane Reforming (SMR), produces a high amount of CO<sub>2</sub> (1300 times more than the product). This creates a need for new technology that can produce hydrogen without hampering the environment. Thermo-Catalytic Decomposition of methane (TCD) is an attractive approach for CO<sub>x</sub>-free hydrogen production. The energy requirement of the TCD process is quite similar to SMR (37.8 kJ/mol H<sub>2</sub> vs. 41.3 kJ/mol H<sub>2</sub>). Furthermore, the TCD process avoids energy-intensive units, i.e., pressure swing adsorption for CO<sub>2</sub> separation, because it produces CO<sub>x</sub>-free H<sub>2</sub> and solid carbon. Different forms of carbon (i.e., CNTs, graphene) can be made by optimizing the shape and size of the metal crystals of the catalyst. In this context, methane decomposition in a fluidized bed reactor was investigated. An iron-based alumina-supported catalyst was prepared for methane decomposition in the fluidized bed reactor. High methane conversion (>80%) and multi-walled bamboo-shaped carbon nanotubes were obtained by the catalyst. Prepared catalysts, as well as CNTs, were characterized by TPR, BET, FE-SEM, HR-TEM, Raman spectroscopy, TGA, etc. Hydrodynamic aspects of the fluidized bed reactor were also taken into consideration.

**Keywords:** fluidized bed reactor; methane decomposition; hydrogen, carbon nanotubes; thermo-catalytic process.

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### **Conflicts of Interest**

The authors declare no conflict of interest.